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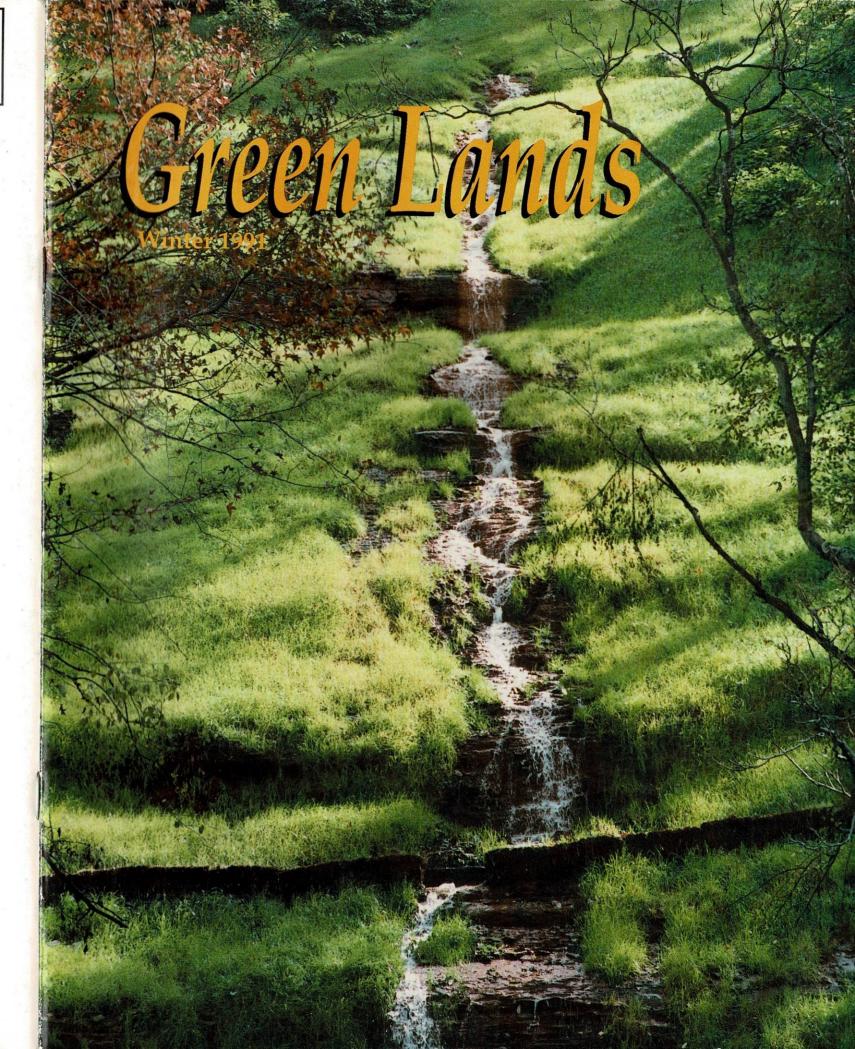
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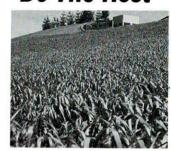
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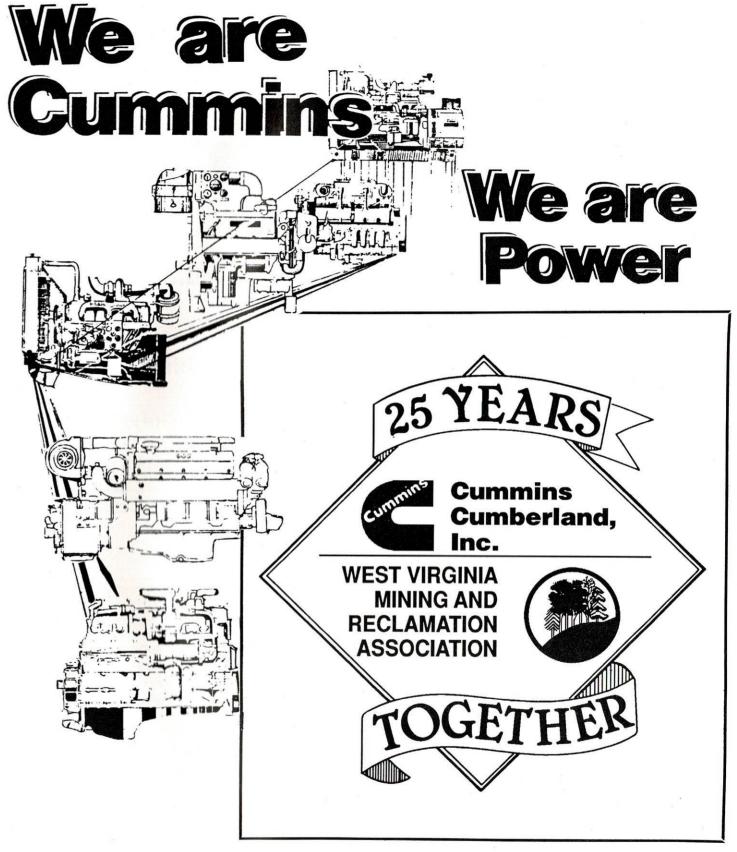
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Green Lands

is a quarterly publication of the West Virginia Mining & Reclamation Association, with offices at 1624 Kanawha Boulevard East Charleston, West Virginia 25311 (304) 346-5318



Our Cover

Ryder & Company created this scene in Fayette County from an abandoned mine site. Now Congress has extended the program under which this feat was accomplished. Our cover story begins on page 20. (Cover photo by Danny Pritt)

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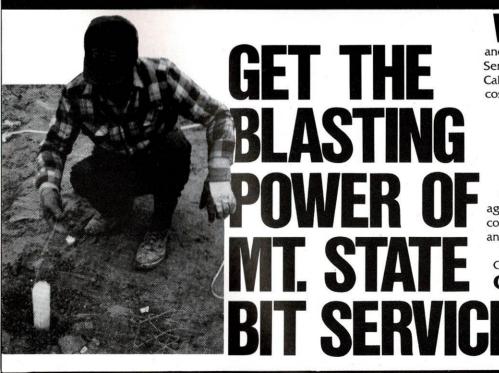
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Fill dirt from the construction site provided material for a contoured bank behind the school.



Site preparation crews also widened an access road.



Site preparation is complete for Barbour County's new vo-tech building extension.

Members Help Out In Barbour County

Several Philippi area companies joined forces late last summer to overcome financial limitations and bring a Barbour County educational expansion to fruition.

Philip Barbour High School opened in 1963, consolidating Barbour County high schools. The Vocational Tech Center was added in 1979. The diverse, modern Vo-Tech Center is served by the Vocation Advisory Council, a collection of business and industry people from surrounding communities.

In turn, it serves the entire county, adults as well as students, providing training in a wide variety of vocational fields. When it was determined by the advisory committee that an expansion of the facility was in order, the ever present question of funding arose.

State funding of \$690,000 was available. "We had State funds to work with, but without the help and cooperation of these companies and people, what we are doing here now just wouldn't have been possible," says Charles Zinn, director of the Vocational Center.

Zinn referred to the site preparation, for which manpower and equipment were donated by several local companies. He estimates a savings of more than \$100,000. "This was

just a tremendous example of public minded cooperation," Zinn added. "The facility will be a source of pride for Barbour County, and so will the spirit of volunteerism that made it possible."

WVMRA members were key players in the project. C&W Coal Co. donated two dozers, a high lift, and surveying services. Stanley Industries, Inc. provided a dozer and a water truck. McCoy Brothers, Inc. contributed a rock truck and a core drill, Grafton Coal Co. came through with a low boy, and 92 Coal Corp. chipped in with a high lift. These same companies also contributed a whole lot of manpower.

The work itself was a model of efficiency, carried out somewhat in the manner of mountaintop removal surface mining. The primary task was to level a bank just behind the existing facility in order to allow the new building to tie into the older one. The fill dirt was used to level out other areas of the school's rolling terrain. An access road was widened, the adjacent bank was contoured, and a new soccer field was constructed.

The entire site preparation project was completed over a three week period in August. Construction on the new 85 x 100 building is scheduled for completion in September.

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'Me? I still drive a truck'

A constant strength of the West Virginia Mining and Reclamation Association through its first 25 years has been the diversity of its membership. Throughout this turbulent period, the organization has been able to call on a great variety of talent, size, and business expertise for advice, support, and encouragement.

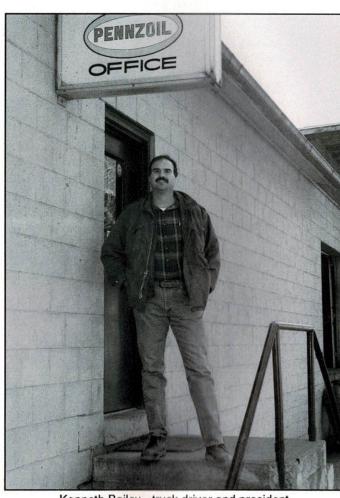
For instance, one of WVMRA's oldest members is a small family owned and run business in Whitesville, West Virginia — B & M Oil Co.

B & M got its start, and its name, in 1963, when Kyle Bailey left the Elk Refining Company to go into partnership with Jim Meadows, the owner of Meadows Oil in Beckley. They formed the B & M Oil Co. In Whitesville, with Meadows as the silent partner, Bailey's wife Nadine as the secretary/bookkeeper, and one truck driver. Though Meadows Oil and B & M were in the same business, and operating in the same general locale, a gentleman's agreement has kept the two company's from stepping on each other's toes for 28 years now.

While he was getting his new company off the ground, Kyle Bailey rented a house about 500 feet down the road from his office. Later he bought the house and there they stayed.

The Baileys retained their relationship with Elk Refining, serving as a distributor for Elk's principal product, Keystone Gasoline. Later, Elk Refining was purchased by Pennzoil, and B & M has remained a Pennzoil distributor to this day, covering southern West Virginia.

In 1968, the Baileys bought Meadows' share of the business, but retained the now familiar name. Nadine became the secretary/treasurer, a post she still holds. In 1974, B & M built a new office facility on the adjacent lot, and converted the old one into a garage.



Kenneth Bailey - truck driver and president

Through the changes, the Bailey's son Ken was growing up in the business. "I started out as a teenager, washing trucks," he recalls. "Later on, I was able to drive them."

And what is his role with the company in 1991. "Well, I'm still driving trucks," he laughs. "Just part-time, though. I'm also the president."

After training at all levels of the company, Ken became its vice president, assuming the mantle of leadership when his father passed away in 1987.

Though he still takes his turn at truck driver when the situation dictates, he has a full-time staff of four drivers and three office people.

B & M Oil has never grown into a huge company in its quarter century plus. But it has provided a good living for a good many folks over those years, and it's the kind of operation that has formed the foundation of WVMRA since the organization's inception 25 years ago this year.



Company founder Kyle Bailey in 1963.



Keystone Oil was the product then, and it's still Pennzoil today.

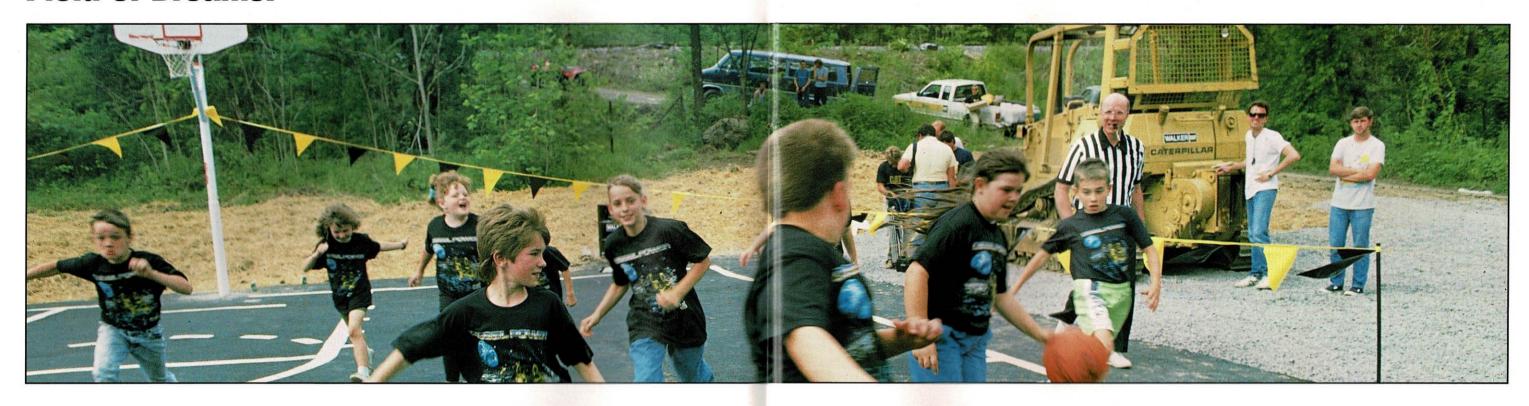
General Paving Co.

Contractors for the Ruthbelle Project in Preston County and the Kennedy Subsidence Project in Upshur County

Congratulations to all concerned with the nation's foremost Abandoned Mine Lands program. With the cooperation of the coal industry, and the professionals within the West Virginia Division of Energy's AML section, West Virginia will continue to lead the nation in reclamation.

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equipment, manpower and materials. • The season opener was a success. A nice crowd was on hand and West Virginia University basketball coach, Gale Catlett, conducted a free basketball clinic for the kids. • Making a difference for the community. That's what it's all about. And that's the Walker Difference.

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In participation with AML projects, we are utilizing the dollars collected from the West Virginia coal industry to make our state more beautiful.

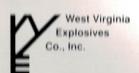
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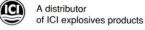


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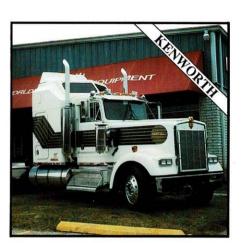
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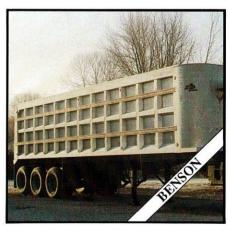
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The "Carbondale Portals & Slide Project" in Fayette County, completed by Ryder & Co. of St. Albans, exemplifies the excellent results of West Virginia's AML program.

photos by Danny Pritt

Congress extends AML program

The Abandoned Mine Reclamation Fund, established by the Surface Mine Control and Reclamation Act of 1977, has been reauthorized by Congress.

Due to expire in 1992, the Fund has been extended through September 1995. The Abandoned Mine Reclamation Act of 1990 contains several other significant provisions.

- Interest may now accrue on the federal share of AML funds. West Virginia coal operators have now contributed more than \$300 million to the AML fund, which took several years to get off the ground. For 13 years, no interest was collected on these accumulated funds, a major potential source of revenue. This provision will enhance the fund, at no cost to operators.
- Substantial additional funding may be available for high priority projects. From the outset, AML has allocated its funds according to priority criteria in an attempt to address the most serious environmental problems first. Fifty percent of the money collected within a state is made available to that state for AML projects. The other half is directed to the (Interior) Secretary's Fund for use in a variety of areas. Hereafter, 40% of the Secretary's Fund will go to a supplemental allocation to the states for comple-

tion of priority 1 & 2 projects. Thus, each state will be eligible to recover a larger share of the reclamation fees collected from its industry.

- Further funds from the secretarial share may be utilized by states to address projects abandoned after August 3, 1977. These sites, mined during the long interim period between federal legislation and enabling regulation, have not previously been included in AML inventories, though they may have the same environmental urgency as older sites.
- Funding will also be made available for the treatment of acid mine drainage. Each state program is eligible to receive an additional amount, equal to 10% of its AML funds, to be used to "protect, repair, replace, construct, or enhance facilities relating to water supplies . . ." Enabling legislation to take advantage of this provision has been introduced in the West Virginia Legislature.

In West Virginia, plans are underway to accelerate the AML program. Director Pete Pitsenbarger has committed his section to contracting one project in every week of 1991. Meeting this goal would build on an already successful program.

In recent years, West Virginia's AML program has



Pioneer Construction - Omar Refuse - Logan County

achieved commendable results, utilizing industry generated funds to provide employment and enhance the environment. One notable example is the "Carbondale Portals & Slide Project" in Fayette County (cover photo), completed by Ryder & Co. of St. Albans.

Located near Smithers, in western Fayette County, the project involved some 2 million gallons of water, which was trapped in an old underground mine, where seepage was causing surface slides. With private residences located just below the mine site, this project was a definite "priority 1"

After accepting the contract in November of 1989, Ryder & Co. successfully drained the mine, removed two slides from the old bench, and completely revegetated the site.

"This project is emblematic of why the AML program was conceived, and why it was re-authorized," says Construction Administrator Danny Pritt. "In a few short months, a significant environmental hazard was eliminated at no cost to the taxpayer."

"Hopefully, projects of this nature will provide the impetus for Congress to extend the AML program well beyond 1995."



photo by Pete Pitsenbarger

Pioneer Construction - Muddy Creek - Preston County







General Paving - Kennedy Subsidence - Upshur County

photo by Danny Pritt

photo by David L. Smith

General Paving - Ruthbelle - Preston County





Kimberly Industries - Sarah Ann - Logan County

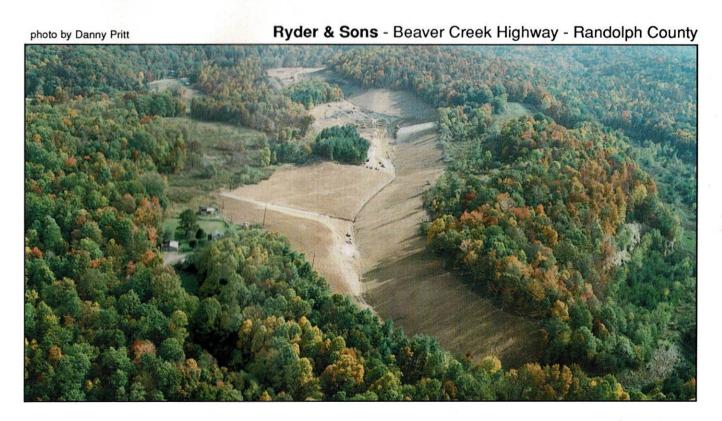
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24 Green Lands



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and to the West Virginia Division of Energy,
for the cooperation and dedication
which has allowed the
AML program to go forward,
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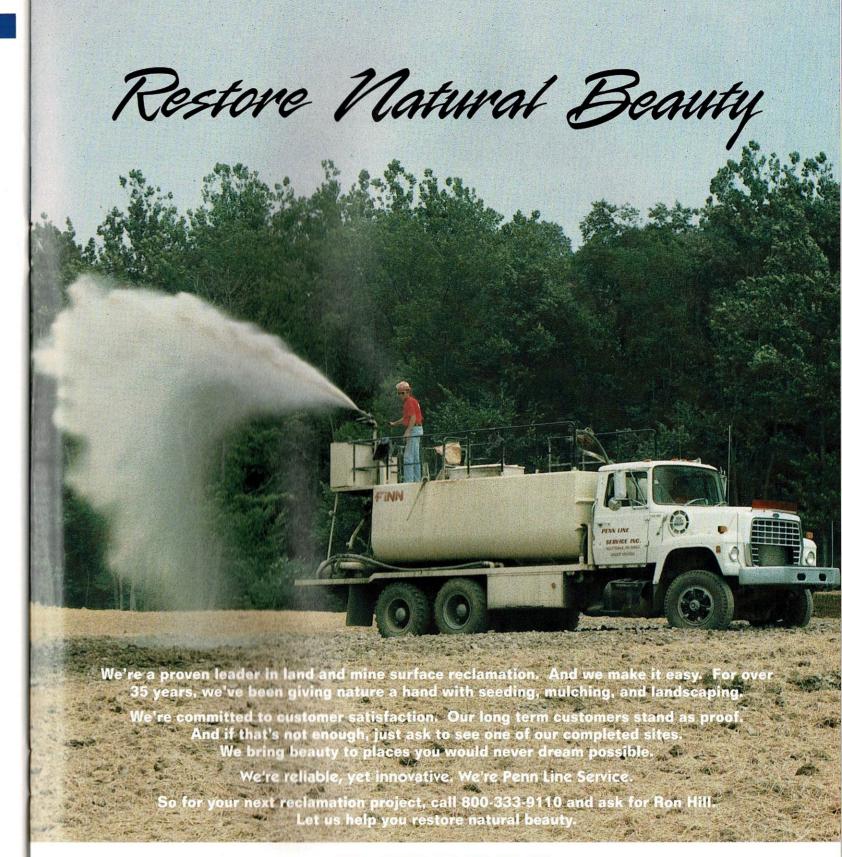
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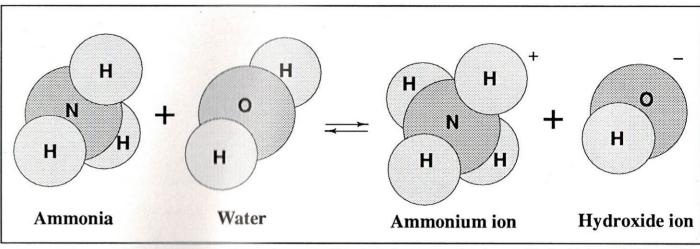


Figure 1 - Ammonia reacts with water to form ammonium ions and hydroxide ions. In this reaction, water acts as an acid (hydrogen donor), and the ammonia acts as a base (hydrogen acceptor).

Using Ammonia to Treat Mine Waters

Ben B. Faulkner
National Mine Land Reclamation Center
Jeff Skousen
West Virginia University

Note: This article is an extract from Handbook For Use of Ammonia in Treating Mine Waters, published by WVMRA in November, 1990.

Introduction

Treating water with anhydrous ammonia is not a new practice, but the advantages of ammonia over other neutralizing agents are so great its popularity is increasing in the mining industry. Along with this increase in use has come a stigma of uncertainties as to whether ammonia is a safe, environmentally sound agent.

WVMRA has responded to the pleas of responsible operators for information and assistance on this technology to prepare a handbook providing this assistance. The 28 page handbook was prepared with the help of leading reclamationists and environmental chemists involved with acid mine drainage in West Virginia.

Acid Mine Drainage Treatment

Coal mining sometimes liberates materials within the coal seam or overburden that produce acidity as they are exposed to moisture and air or dissolved oxygen. Pyrite and associated minerals in coal and overburden contain sulfides and metals like iron, manganese, and aluminum. If hydrologic conditions allow the flow of water through these acid generating areas within the mine or refuse area, the resultant water emanating from the backfill, deep mine, or refuse area is acid mine drainage.

In an attempt to protect aquatic life and water uses further downstream, coal operators use various techniques to prevent, ameliorate, or neutralize the acidic drainages. Operators use anhydrous ammonia for the\neutralization of acid mine drainage and water treatment in coal preparation plants.

When ammonia is introduced to an acid solution, some of the ammonia reacts directly with the acid, consuming acidity, raising pH, and producing the ammonium ion. Introduced ammonia may also associate itself with several water molecules without undergoing an actual chemical conversion. This hydrated ammonia is commonly referred to as unionized ammonia and is toxic to aquatic life forms at low concentrations.

Simultaneous with other reactions, ammonia may also react directly with water, producing ammonium and hydroxyl (base) ions (Figure 1).



Figure 2 - Ammonia, when injected into acid mine drainage, reacts rapidly to raise pH and to neutralize acidity. Metals dissolved in the water will precipitate as the pH is raised.

The ammonium and hydroxyl ions produced will react with sulfuric acid from mine drainage to produce ammonium sulfate and water. Acidity is consumed and pH is increased. Ammonium sulfate is somewhat soluble in water and acid solutions; a portion may remain dissolved, while some may precipitate (drop out in the treatment pond sludge). Hydroxyl (base) ions will react with dissolved iron to form an insoluble compound, iron hydroxide. Thus, in addition to raising pH, ammonia treatment can also remove dissolved metals from mine drainage (Figure 2).

Once anhydrous ammonia is delivered to the minesite, there is little chance for accident or injury. Ammonia has been safely used in industry and agriculture for many years. Anhydrous ammonia is, however, a hazardous, pressurized gas, which deserves respect.

Dry ammonia, which contains no water, is referred to as anhydrous ammonia. It is a pungent, colorless gas which may be compressed and cooled to a colorless liquid. It is lighter than air, and is extremely soluble in water. Ammonia has been classified by the USDOT as a non-flammable

compressed gas. Having a very narrow flammable range in air, its products of combustion are neither toxic or hazardous.

There are several operational advantages to ammonia for neutralization of acid. The cost is low and stable. Ammonia will not freeze, gel, or solidify at low temperatures so it may be used in all seasons. As a compressed gas, it does not have to be pumped or gravity fed. A storage tank may be located, based on convenience, and not limited by topography or the availability of power. It is homogeneous, and is sufficiently soluble that no additional mixing with the acid solution is necessary. There are no heavy bags or drums to transport or store. Once a tank is in place and properly plumbed, ammonia is delivered and dispensed by means of valves. Tanks may be rented, and the necessary installation costs are only slightly higher than an equivalent caustic system. The reduction in man-hours which is made possible by using ammonia can be quite significant. The rapid and complete reaction with acidic water makes accurate feed control possible resulting in minimal waste.

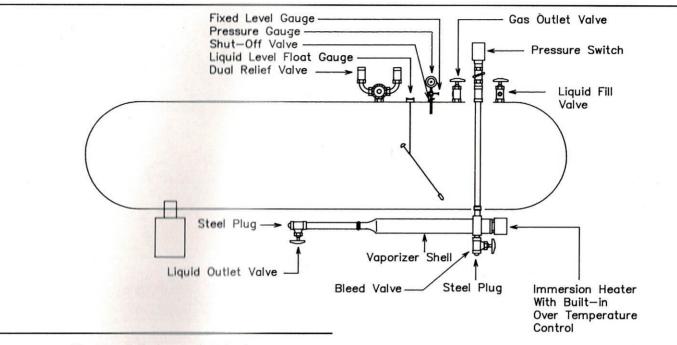


Figure 3 - Typical ammonia tank setup.

Environmental Effects of Ammonia Human Exposure to Ammonia

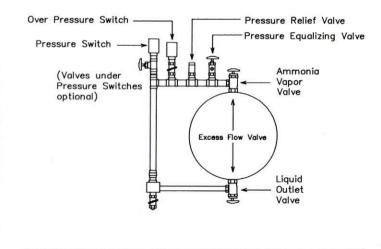
Ammonia, strictly speaking, is not a poison. Ammonia is produced by almost all living creatures; thus it has no cumulative, toxic effect, as lead or mercury have. It does not cause cancer. However, in concentrated form, it has a powerful corrosive action on human tissue. Fortunately, ammonia's pungent odor is detectable at levels lower than what is considered dangerous. It is unlikely an individual would be exposed to levels of ammonia higher than allowed by law.

Only occasional maintenance is necessary on a well-installed ammonia system. All piping, joints, tank, and tank fittings should be examined periodically for signs of corrosion, damage, wear, and leaks (Figure 3).

Leaks in ammonia pipe can usually be detected by odor. All leaks should be approached with caution and with full protective gear (gas masks and gloves with protective clothing). Even a very small leak can be located with sulfur tapers or moist litmus paper.

Regulations and Reporting

United States Department of Labor regulations as set forth in 29 CFR 1910.1000 state; "An employee's exposure to ammonia must be limited to a concentration not to exceed 25 ppm of ammonia in air by volume based upon an 8 hr. time weighted average".



Anhydrous ammonia is reportable under the S.A.R.A. Title III Regulations - "Community Right to Know Act". The purpose of this act is of a twofold nature; to establish a program to provide the public information on hazardous substances and toxic chemicals in the community, and to establish Emergency Response Plans and notification requirements to protect the public. To satisfy the purpose of these regulations, users of anhydrous ammonia must annually submit information to several local and state emergency response organizations.

Environmental Compliance

The use of anhydrous ammonia requires a modification of the operator's NPDES permit. The first step an operator interested in using anhydrous ammonia should take is to analyze the receiving stream for existing total ammonia, nitrogen, nitrites, nitrates, alkalinity, acidity, pH and sulfates. It is possible the operation may be located in an area where the existing ammonia content is high already as a result of sewage, farm drainage, hatchery effluent, or other influence. If the background un-ionized ammonia is at or above the respective stream criteria, the modification will likely be denied. It is strongly suggested that a biological survey be conducted in the receiving stream to document macroinvertebrate diversity prior to the introduction of ammonia. A biological survey is usually limited to the benthics (mostly insects which inhabit the bottom of the stream) which generally exhibit stress from pollution before higher aquatic life.

If the modification is approved, the operator must monitor his discharge and must not cause the receiving stream to exceed the numerical criteria of 0.05 mg/l of un-ionized ammonia (0.02 mg/l if classified as a a trout stream).

Biology of Anhydrous Ammonia

Although anhydrous ammonia is a commercial chemical product purchased through a chemical supplier, it is important to realize that its active ion, ammonium, is a natural substance produced by fish and other organisms, including humans, and is released from the body in waste material. It is an integral element in the basic bio-cycles of matter and energy. Ammonia is a biologically active compound present in most waters as a normal degradation product of nitrogenous organic matter (Figure 4).

Several factors have been shown to modify acute NH₃ toxicity in fresh water. Some factors alter the concentration of un-ionized ammonia by affecting the equilibrium of ammonia, and others ameliorate or exacerbate its effects. These factors include dissolved oxygen concentration, temperature, pH, previous acclimation to ammonia, fluctuating or intermittent exposures, carbon dioxide concentration, salinity, and the presence of other toxicants.

Since an increase in temperature and pH will increase the portion of un-ionized ammonia, the summer and fall months can be critical. During droughts, surface runoff is less available to dilute the discharges from fairly steady or belated seepage from surface mines or deep mines. These discharges generally constitute a larger percentage of the stream flow during dry times than during normal flow conditions. Lesser volumes or less buffered surface waters will have a diminished ameliorative effect on either acid mine drainages or over-treated drainages.

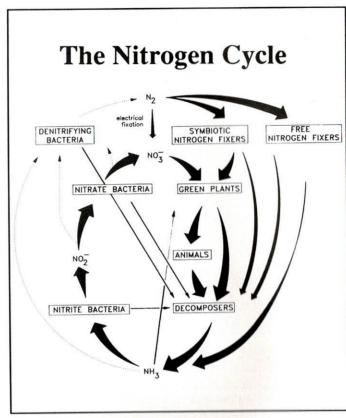
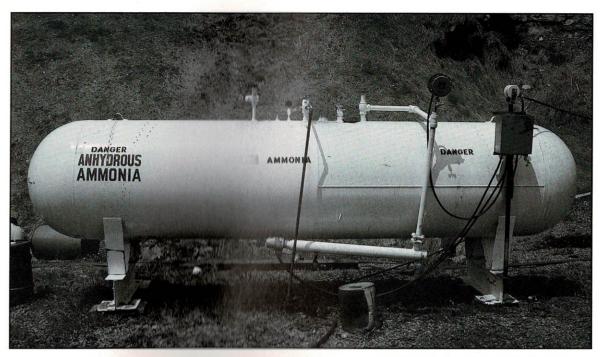


Figure 4 - A graphic representation of the nitrogen cycle. Nitrogen can exist in eight different forms (from NO₃ to NH₄+), and converts readily between forms.

Eutrophication

Excessive nutrients, such as nitrates and phosphates commonly originate in domestic sewage, run-off from agricultural fertilizer, waste materials from animal feed lots, packing plants, etc. These nutrients cause pollution primarily because they stimulate the growth of microorganisms which often increase the biological oxygen demand (BOD) of the water and reduce the amount of dissolved oxygen available for fish and higher animals.

Even the most modern secondary sewage plants, which precipitate solids and inactivate most bacteria in domestic wastewater, do not remove the basic nutrients such as ammonia nitrogen, nitrates, nitrites, and phosphates. These nutrients stimulate algal growth and lead to plankton blooms, producing obnoxious tastes and odors in water, excessive plant growth causing stagnation and disruption of aquatic ecology. These conditions occur only when more nutrients are introduced into the system than it can handle. Many natural waters are nitrogen limited in their productivity, and small additions can actually be beneficial.



Acidification

It has been suggested that nitrification can occur downstream from a treatment pond and ultimately result in the regeneration of acidity and objectionable levels of nitrates. The bacteria involved in nitrification have an optimal range of growth conditions. The rate of bacterial growth and hence the rate of nitrification is both temperature and pH dependent. In most streams, this oxidation reaction would occur very slowly if it occurred at all.

Obviously, ammonia is a natural substance that is involved in a number of natural processes. The introduction of ammonia into a watercourse for the neutralization of AMD or other coal preparation purposes will enter these cycles and will cause problems downstream only when conditions or concentrations exceed the system's ability to handle the ammonia. Conditions for nitrification are not optimum in most downstream situations. It is more likely that ammonia will undergo further dilution, volatilization, and assimilation prior to entering a nitrification prone environment. Similarly, most ammonia rich sewage discharges are not acidic in nature.

Nitrogen exists in the environment in many oxidation states. Microorganisms interconvert these forms of nitrogen by various oxidative or reductive pathways. Thus, there exists within the environment the potential to remove excess ammonium as gaseous nitrogen. If this capacity did not exist, the runoff from the 12.5 million tons of ammonia used annually in American agriculture would have acidified and eutrophied all our surface waters.

Figure 5 - An ammonia tank equipped with a vaporizer. The vaporizer is a heating element which helps maintain a consistent pressure in the tank for uniform vapor dispensing.

Field Applications of Ammonia Surface Water Applications

Ammonia is not an ideal chemical for all water treatment applications. While it is usually less expensive than caustic soda or soda ash, it may be more expensive than lime products at a capital intensive long-term treatment facility. Each treatment situation is unique with characteristics that will influence chemical and delivery mechanism selection (see *Green Lands* - Fall 1990, pp 31-37, "Acid Mine Drainage Treatment Systems: Chemicals & Costs").

The target pH to achieve effluent limitations is dependent on the type and concentration of metals present in the water, and the available retention time for sedimentation. If manganese is present in the influent water, the pH will have to be raised to approximately 9.8 with ammonia, or to 10.0-10.4 with caustic to precipitate manganese. This would require far more reagent than treating water to remove only iron at pH 8 to 9. It is strongly suggested that bench scale titration or neutralization of the water to be treated be conducted to determine the amount of various reagents necessary to meet effluent limits. The volume and characteristics of the precipitant resulting from the bench tests will indicate the relative difficulty in sludge handling.

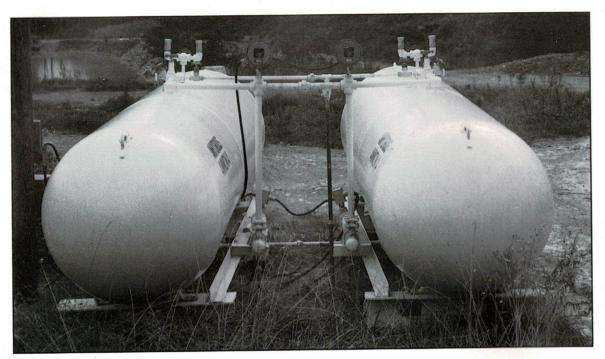


Figure 6 - Small ammonia tanks may be plumbed together to increase internal surface area of the tanks, which improves vapor pressure.

Vapor Dispensing

Anhydrous ammonia may be dispensed either as a liquid or vapor. Vapor is far easier to regulate than liquid. Initially, coal operators used liquid ammonia to batch treat impoundments, sometimes controlling the submerged injector with a solenoid valve tied to a pH probe further downstream. Ammonia is extremely soluble in water but may stratify within the pond when applied in this manner. Additionally, flow rates from the injector site to the pH sensor may be so slow that wide variation in pH occurs due to this response time. Further complicating this treatment arrangement is the influence of varying pressure in the lines due to temperature variation. Many of these operational difficulties can be overcome by dispensing vapor ammonia on a continuous basis at the entrance channel(s) of the impoundment.

Further, it is easier to treat the influent water to a pond in a continuous fashion rather than batch treat a pond. Steady flows with consistent water chemistry can be treated with a minimum of attention, but will still have to be monitored regularly. If line power is accessible, the addition of a vaporizer will greatly aid in regulation during cold weather months (Figures 5 & 6).

Coal Preparation Plants

Ammonia can be used in preparation facilities safely and without odor. Ammonia will not adversely affect the flocculants used in the plant and the gelatinous precipitates will assist in clearing the fines. Ammonia can be added as a vapor or as a liquid when rapid neutralization is desired. This flexibility is a great advantage for plants which wash several seams of coal which have different effects on plant pH. Use of ammonia in preparation plants differs from exterior ammonia treatment installations primarily in that within the plant you are working in a respirable atmosphere without the free flow of air or wind to dilute or remove the ammonia vapor. Additional ventilation in critical areas may be necessary if monitors indicate high levels of ammonia.

Ammonia is a strong reagent. Like most industrial chemicals, it has the capacity to alter the environment into which it is introduced. Unlike many substances, it is a natural, abundant compound which nature can easily handle in moderate amounts. It is the responsibility of the ammonia user to see that those limits are not exceeded by dispensing it in a safe, environmentally sound manner.

Acknowledgements

Thanks to Tiff Hilton, Martha Hilton, Ron Lilly, and Jim Hajec. The Handbook For Use of Ammonia in Treating Mine Waters, published November, 1990 by the West Virginia Mining & Reclamation Association, Charleston, WV. This publication is avialable, at a cost of \$5, From WVMRA, 1624 Kanawha Blvd. E., Charleston, WV 25311.

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Coal Calendar

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- 9-11 American Mining Congress/National Coal Association, Fourth Communications Workshop, Westin-Tabor Center, Denver, CO, contact Carol Sheppard, Workshop Coordinator, AMC, 1920 N St. NW, Washington, D.C. 20036, (202) 861-2839.
- 9-11 Short Course, "Conveyor Belt Network Design," University of Kentucky, Lexington, KY, contact Mary Lou Johnson, Institute for Mining and Minerals Research, 411 Breckinridge Hall, Lexington, KY 40506, (606) 257-2846.

April

- 3-4 West Virginia Surface Mine Drainage Task Force Symposium, Ramada Inn, Morgantown, contact Patty Bruce, West Virginia Mining & Reclamation Association, 1624 Kanawha Blvd. E., Charleston 25311, (304) 346-5318.
- 9-11 1991 Mine Health, Safety & Training Conference, Charleston Town Center Marriott, Charleston, contact Tony Grbac, West Virginia Department of Energy, 1615 Washington St. E., Charleston 25311, (304) 348-3500
- Due date for papers for the Pennsylvania Blasting Conference (November 14-15, 1991), submit abstracts to Chuck Herd, Conference Coordinator, J. Orvis Keller Bldg., Penn State University, University Park, PA 16802, (814) 865-3003.
- 30-2 Coal Prep 91, Lexington, KY, contact Sam Posa, Maclean Hunter Presentations, 12371 E. Cornell Ave., Aurora, CO 80014, (303) 696-6100.

May

- 12-16 1991 National Meeting of the American Society for Surface Mining and Reclamation, Durango, CO, contact Bill Plass, ASSMR, 21 Grandview Dr., Princeton, WV 24740, (304) 425-8332.
- **Short Course, "Blast Casting,"** West Virginia University, contact Sherry Strahin, Department of Mining Engineering, P.O. Box 6070, Morgantown, (304) 293-7680, FAX 293-5708.
- 21-23 Short Course, "Surface Mine Blasting," West Virginia University, contact Sherry Strahin, Department of Mining Engineering, P.O. Box 6070, Morgantown, (304) 293-7680, FAX 293-5708.

June

- **5th U.S. Mine Ventilation Symposium**, West Virginia University, contact Sherry Strahin, Department of Mining Engineering, P.O. Box 6070, Morgantown, (304) 293-7680, FAX 293-5708.
- 10-12 10th conference on Ground Control in Mining, West Virginia University, contact Sherry Strahin, Department of Mining Engineering, P.O. Box 6070, Morgantown, (304) 293-7680, FAX 293-5708.

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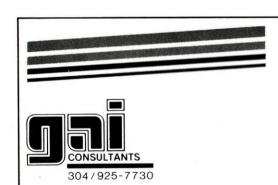
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